THE FACILE (<-50°C) FORMATION OF <u>cis</u>-BICYCLO[6.1.0]NONA-2,4,6-TRIENES FROM CYCLONONA-1,3,5,7-TETRAENES: A SYMMETRY - FORBIDDEN REACTION ?

G.Boche*, M.Bernheim, D.Lawaldt and B.Ruisinger

Institut für Organische Chemie der Universität München, Karlstraße 23, D-8000 München 2

<u>Summary</u>: The formation of the <u>cis</u>-bicyclo[6.1.0]nona-2,4,6-trienes <u>4a-c</u> is not due to the sequence $1 \rightarrow 2 \rightarrow 4$, which would involve the symmetry-forbidden reaction $2 \rightarrow 4$ even at -50° C. Rather, reaction of RX at C⁴⁻⁷ of 5, which is formed together with <u>1</u>, leads (probably via <u>6a-c</u> and <u>7a-c</u>) to <u>4a-c</u>.

Thermally symmetry-forbidden electrocyclic reactions ¹⁾ are normally reluctant to occur at any measurable rate below 0°C²⁾. Thus, the facile (<50°C) formation of substituted <u>cis</u>-bicyclo[6.1.0]nona-2,4,6-trienes (e.g. <u>4c</u>) from <u>cis</u>,-<u>cis</u>,<u>cis</u>,<u>cis</u>-cyclonona-1,3,5,7-tetraenes (e.g. <u>2c</u>)³⁾, which seems to hurt the Woodward-Hoffmann rules ³⁾, would be quite exceptional. The properties and the protonation reaction of the all-<u>cis</u>-[9]annulene anion (<u>1</u>) and its <u>trans</u>-isomer 5^{4} , the electrophilic amination ⁵⁾ of these anions and a reexamination of the origin of <u>4c</u>, however, lead to the conclusion that <u>4a-c</u> are not formed via a forbidden reaction from <u>2a-c</u>. Rather, the genesis of <u>4</u> traces back to some <u>trans</u>-anion <u>5</u>, as shown below.

If all-<u>cis</u>-anion <u>1</u> is reacted with electrophiles RX, only all-<u>cis</u>-2 can arise (Fig.1.). Thus, the protonation of <u>1</u>⁶⁾ gives <u>2a</u>, which isomerizes at 35°C ($\mathcal{T}_{1/2}=16 \text{ min}$) to <u>3a</u>⁷⁾. Likewise, the amination of <u>1</u>⁶⁾ with N,N-dimethyl-O-(oxo-diphenoxy-phosphoryl)-hydroxyl amine ⁸⁾ in tetrahydrofuran (THF) between -30 and 0°C, followed by workup (2n HCl, 2n NaOH) at 0°C, leads to <u>2b</u> (¹H-nmr (CDCl₃): $\delta = 2.27$ (s, 6H, N(CH₃)₂), 3.86 (m, 1H, H¹), 5.17-6.10 (m, 8H, H²-H⁹)). On warming to 30°C, <u>2b</u> isomerizes to <u>3b</u>⁹. A similar behavior is also observed if <u>1</u>⁶ is alkylated with α -bromobenzyl acetate ³⁾ in THF between -50 and 0°C: <u>4c</u>³ is not detectable ¹⁰, as this is the case with <u>4a</u> in the protonation and <u>4b</u> in the amination reaction of <u>1</u>⁶.

The reaction of the <u>trans</u>-anion 5, which is formed in the reaction of $9-\frac{\text{anti}}{3i^{4}a_{p}}$, chloro-<u>cis</u>-bicyclo[6.1.0]nona-2,4,6-triene with lithium in addition to $1^{3i^{4}a_{p}}$, with electrophiles RX is also shown in Fig.1: depending on the site of the reaction not only all-<u>cis</u>-cyclononatetraenes 2 but also <u>cis</u>, <u>trans</u>, <u>cis</u>, <u>cis</u>, <u>and</u>/ or <u>trans</u>, <u>cis</u>, <u>cis</u>-isomers 6 and 8, respectively, can be formed. This is clearly demonstrated by the protonation of $5^{4a,11}$: <u>6a</u> and <u>8a</u> are trapped with tetracyanoethylene between -50 and 60° C and they isomerize already at around -30° C to <u>4a</u> (!) and <u>9a</u>, respectively; <u>2a</u> leads to <u>3a</u> at 30° C. Similar-

ly, in the amination of 5 between -30 and 0° C, followed by workup at 0° C, one observes 10 the amine 4b 9 in $40\pm5\%$ yield. Again, the alkylation of 5 with α -bromobenzyl acetate 3) in THF between -50 and 0°C is in accord with the protonation and amination reactions of 5: 4c³⁾ is formed ¹⁰⁾ in 40±5% yield. Therefore, in analogy to $\underline{4a}$, $\underline{4b}$ and $\underline{4c}$ should result from $\underline{6b}$ and $\underline{6c}$, respectively, which originate from the amination $\binom{8}{}$ and alkylation $\binom{3}{}$ at $\binom{4-7}{}$ of 5.

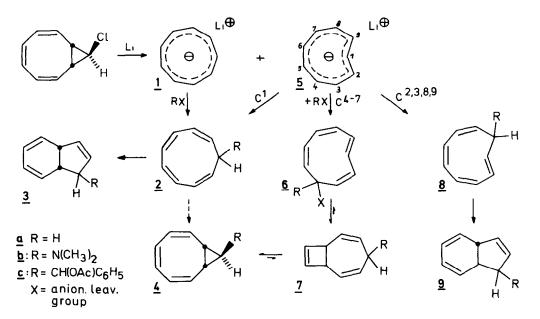


Fig.1.: Reactions of anions 1 and 5 with electrophiles RX.

References and notes

- 1) R.B.Woodward and R.Hoffmann, J.Am.Chem.Soc. <u>87</u>, 395 (1965).
- 2) a. J.E.Baldwin, "Thermally Forbidden Reactions", in A.P.Marchand and R.E.Lehr, "Pericyclic Reactions", Vol.II, Academic Press, New York, San Francisco, London 1977, p.273; b. H.Prinzbach, H.Babsch and D.Hunkler, Tetrahedron Lett. 1978, 649.
- 3) a. M.Neuenschwander, W.Rutsch and P.Engel, Tetrahedron Lett. <u>1979</u>, 233; b. W.Rutsch, A.Frey, M.Neuenschwander and P.Engel, Helv.Chim.Acta 62, 718 (1979).
- 4) a. G.Boche, H.Weber, D.Martens and A.Bieberbach, Chem.Ber. <u>111</u>, 2480 (1979); b. G.Boche and A.Bieberbach, Chem.Ber. <u>111</u>, 2850 (1978).
- 5) G.Boche, N.Mayer, M.Bernheim and K.Wagner, Angew.Chem. <u>90</u>, 733 (1978).
- The ¹H-nmr spectrum reveals $\ge 96\%$ of <u>1</u> and $\le 4\%$ of 6 7) G.Boche, D.Martens and H.Böhme, Angew.Chem. <u>81</u>, 565 (1969); P.Radlick and G.Alford, J.Am.Chem.Soc. <u>91</u>, 6529 (1969); A.G.Anastassiou, V.Orfanos and J.H.Gebrian, Tetrahedron Lett. <u>1969</u>, 4491; S.Masamune, P.M.Baker and K. Hais.
- K.Hojo, Chem.Commun. <u>1969</u>, 1203.
- 8) G.Boche and M.Bernheim, unpublished results.
- 9) J.M.Brown and M.Ogilvy, J.Am.Chem.Soc. <u>96</u>, 292 (1974).
- 10) The cyclopropyl hydrogens of $\frac{4a-c}{a-c}$ are easily seen in the ¹H-nmr spectrum. 11) G.Boche, H.Weber and J.Benz, Angew.Chem. <u>86</u>, 238 (1974).

Acknowledgment: Support from the Deutsche Forschungsgemeinschaft and the Fonds der Chemischen Industrie is gratefully acknowledged. Cyclooctatetraene was kindly provided from the BASF AG, Ludwigshafen.

(Received in Germany 28 August 1979)